20

35

(2.94~g,~1.05~eq. based on MeSO $_3$ H) and concentrated in vacuo. To the resulting residue was added isopropanol (120 mL) and the mixture was concentrated in vacuo to give 88 g of a residue.

The residue was dissolved in isopropanol (37 g) at 47° C. 5 The resulting solution was cooled down to 2° C. during 2.5 h; crystallization started spontaneously at 30° C. The crystalline product was isolated by filtration, washed with isopropanol (3×20 mL, 0° C.) and dried in vacuo (17 h at 35° C.) to give 10.1 g of a white crystalline product which according to GC 10 consisted of 96.4 wt % α -(4) and 0.065 wt % β -(4), corresponding with a total yield based on E-R-3-(2,2-dimethyl-[1, 3]dioxolan-4-yl)-acrylic acid ethylester of 9% and an α -(4): β -(4) ratio of >1000:1.

Thus, the total yield of the first and second crop of α -(4) 15 based on E-R-3-(2,2-dimethyl-[1,3]dioxolan-4-yl)-acrylic acid ethylester was 46%.

Example 8

Preparation of pure (3R,3aS,6aR) hexahydro-furo[2,3-b]furan-3-ol from α -(4) intermediate

The procedure described in WO03/022853, Example IV, $_{25}$ last step, was followed.

Example 9

Preparation of pure α -(4) from R-3-(2,2-dimethyl-[1, 3]dioxolan-4-yl)-acrylic acid ethylester by Direct Crystallization of α -(4) from a Crude Mixture of β -(4) and α -(4) and Simultaneous Epimerization of β -(4) to α -(4)

To R-3-(2,2-dimethyl-[1,3]dioxolan-4-yl)-acrylic acid ethylester (399.5 g, 75.1 wt % pure, 1.5 mol) was added nitromethane (915.0 g of a 11 wt % solution in methanol, 1.65 mol, 1.1 eq.) and the solution was cooled to 0° C. Subsequently, DBU (233.3 g, 1.5 mol, 1 eq.) was added dropwise 40 during 50 min at 0-5° C. and the reaction mixture was heated up to 20° C. and stirred for another 16 h at that temperature. The resulting reaction mixture was cooled to 0° C. and NaOMe (594.0 g of a 15 wt % solution in methanol, 1.65 mol, 1.1 eq.) was added dropwise during 50 min at 0° C. The 45 resulting solution was stirred for 1 h at 0° C. and quenched into a solution of H_2SO_4 (368 g, 96 wt %, 3.6 mol, 2.4 eq.) in methanol (370 g) at 0-5° C. by dropwise addition during 3 h under vigorous stirring. The reaction mixture was stirred for 2 h at 0-5° C. and then quenched into a stirred slurry of 50 KHCO₂ (457.6 g), in water (870 mL) at 0-5° C. by dropwise addition during 1 h. KHCO₃ was portionwise added to keep the pH above 3.5. The formed salts were removed by filtration at 0-5° C. and washed with methanol (530 mL). After concentration in vacuo of the combined filtrate and washing to 55 approximately 1000 ml the aqueous phase was extracted with toluene (2×2100 mL, 3×1050 mL). The combined organic phases were concentrated in vacuo, giving 202.9 g of a semi-

Subsequently, methanol (42.6 g) and MeSO₃H (6.06 g, 60 0.04 eq.) were added and the mixture was heated up to 50° C. After 2 h of stirring at this temperature the mixture was cooled to 20° C. and stirring was continued for an additional 12 h. After cooling to -5° C., triethyl amine (6.60 g, 1.1 eq. based on MeSO₃H) was added and the mixture was stirred for 65 another 2 h. The crystalline α -(4) which was isolated by filtration, washed with cold (-5° C.) isopropanol (3×70 mL)

and dried on the air. This gave 120.0 g $\alpha\text{-}(4)$ which, according to quantitative GC analysis, was 99.0 wt % pure and contained 0.09 area % of $\beta\text{-}(4)$. This corresponds to a yield of 51% based on R-3-(2,2-dimethyl-[1,3]dioxolan-4-yl)-acrylic acid ethylester.

The invention claimed is:

1. A method for the synthesis of (3R,3aS,6aR) hexahydrofuro[2,3-b]furan-3-ol having the structure of formula (6),

which method comprises the step of reducing the intermediate of formula α -(4):

- 2. A method according to claim 1 which method further comprises crystallizing intermediate of formula α -(4) with a solvent prior to the reduction thereof.
- 3. A method according to claim 1 which method further comprises
 - a) epimerizing with acid intermediate of formula β -(4) into the intermediate of formula α -(4); and

$$O_{\text{MoMe}}$$

$$O_{\text{OMe}}$$

$$O_{\text{OMe}}$$

$$O_{\text{OMe}}$$

$$O_{\text{OMe}}$$

$$O_{\text{OMe}}$$

- b) crystallizing intermediate of formula α -(4) with a solvent prior to the reduction thereof.
- **4.** A method according to claim **3** which method further comprises after crystallizing intermediate of formula α -(4),
 - a) epimerizing with acid intermediate of formula β -(4) in the mother liquor of said crystallization into the intermediate of formula α -(4); and

$$O_{\text{M}} O_{\text{Me}}$$

$$O_{\text{M}} O_{\text{Me}}$$

$$O_{\text{M}} O_{\text{Me}}$$

$$O_{\text{Me}} O_{\text{Me}}$$

$$O_{\text{Me}} O_{\text{Me}}$$